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## Chemical constituents of Nouelia insignis Franch

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Two new diterpenes and *ent*-15 $\alpha$ -hydroxykaur-16-*en*-19-oic acid 11,12-acetonide (**3**), together with 23 known compounds were isolated from the dried aerial parts of *Nouelia insignis* Franch. The structures of new compounds were determined to be *ent*-14 $\beta$ ,15 $\alpha$ -dihydroxykaur-16-*en*-19-oic acid (**1**), *ent*-14 $\beta$ -hydroxy-15-oxokaur-16-*en*-19-oic acid (**2**) on the basis of spectral and chemical evidence. The structure of *ent*-11 $\alpha$ ,16 $\alpha$ -epoxy-15 $\alpha$ -hydroxy-16*S*-kaur-19-oic acid (**4**) was confirmed by X-ray crystallographic analysis.

Keywords: Nouelia insignis; Ent-kaurene; Diterpenic acid; Sesquiterpene lactone; Steroid

#### 1. Introduction

*Nouelia insignis* Franch, an endemic plant and a monotype species of the genus *Nouelia* Franch, is distributed in Sichuan–Yunnan region, especially in Panzhihua City of China. The genera *Nouelia* and *Leucomeris* (Compositae) are cytologically related. *Nouelia* may be descendant of old plants, which was developed as an ornamental plant in China. The plant of *Leucomeris* Kurz, a sibling genus of *Nouelia* Franch, was used as anti-inflammatory medicinal herb to cure cough and snake bite [1]. The chemical investigation on *N. insignis* has not been reported.

This study on the aqueous ethanolic extract of the aerial parts of *N. insignis* led to the isolation of two new *ent*-kaurene-type diterpenes, *ent*-14 $\beta$ ,15 $\alpha$ -dihydroxykaur-16-*en*-19-oic acid (1) and *ent*-14 $\beta$ -hydroxy-15-oxokaur-16-*en*-19-oic acid (2), *ent*-15 $\alpha$ -hydroxykaur-16-*en*-19-oic acid 11,12-acetonide (3) (figure 1), and 23 known compounds *ent*-11 $\alpha$ ,16 $\alpha$ -epoxy-15 $\alpha$ -hydroxy-16S-kaur-19-oic acid (4) [2], ent-15 $\alpha$ -hydroxykaur-16-*en*-19-oic acid (5) [3], *ent*-11 $\alpha$ ,12 $\alpha$ ,15 $\alpha$ -trihydroxykaur-16-*en*-19-oic acid (6) [4], *ent*-12 $\alpha$ ,15 $\alpha$ -dihydroxykaur-16-*en*-19-oic acid (7) [2], 8 $\beta$ ,9-dihydro-onoseriolide (8) [5], taraxerone (9) [6],  $\beta$ -taraxerol (10) [6],  $\alpha$ -taraxerol (11) [7], 2,6-dimethoxy 1,4-benzoquinone (12) [8], stigmasterol (13) [9], 3-*O*- $\beta$ -D-glucopyranosyl stigmasterol (14) [10],  $\beta$ -sitosterol (15),  $\beta$ -daucosterol (16),

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Figure 1. The structures of 1-4.

2,3-dihydroxypropyl hexadecoate (17) [11], palmitic acid (18) [12], tetracosanoic acid (19) [13], pentacosan-1-ol (20) [14], taraxasteryl acetate (21) [15,16], pseudo taraxasteryl acetate (22) [17], caffeic acid (23) [18], sucrose (24) [19], glutinol (25) [20], and rutin (26). These compounds were identified prodominantly on the basis of spectral data. The structure of 4 was confirmed by X-ray crystallographic analysis (figure 3).

#### 2. Results and discussion

Compound 1 was obtained as colourless needles. Two tertiary methyls ( $\delta_{\rm H}$  0.80, 3H, s;  $\delta_{\rm H}$ 1.12, 3H, s;  $\delta_{\rm C}$  15.6, 29.0), an *exo*-methylene ( $\delta_{\rm H}$  4.91 brs, 4.89 brs;  $\delta_{\rm C}$  105.7), and two oxygenated methines ( $\delta_{\rm H}$  3.58 d, 3.76 s;  $\delta_{\rm C}$  64.9, 81.9) were recognized from the <sup>1</sup>H and <sup>13</sup>C NMR spectral data (table 1). Twenty carbon signals were observed in the <sup>13</sup>C NMR spectrum, which were in accordance with its molecular formula  $C_{20}H_{30}O_4$ , provided by the quasi-molecular ion peak at m/z 357.2043 [M + Na]<sup>+</sup> in the HRESIMS. Comparison of the evidence mentioned above and the NMR spectral data in table 1 with those of ent- $12\alpha$ ,  $15\alpha$ -dihydroxykaur-16-en-19-oic acid, isolated from Pteris longipes [2], suggested that compound 1 should be also an *ent*-kaurane type diterpene. The IR absorption bands at  $v_{max}$ 3430, 3427, 3200–2600, and  $1694 \text{ cm}^{-1}$  showed the presence of hydroxyl groups and the carboxyl group, therefore according to the molecular formula, two hydroxyl groups and one carboxyl group. The assignment of the <sup>1</sup>H and corresponding <sup>13</sup>C NMR signals was succeeded on the basis of HMQC and  ${}^{1}H^{-1}HCOSY$  experiments. Two hydroxyl groups were located at C-14 and C-15 based on the HMBC correlations (figure 2) from H-17 ( $\delta$  4.89 and 4.91, each 1H, brs) and H-9 ( $\delta$  1.41, 1H, s) to C-15 ( $\delta$  81.9), and from H-9 ( $\delta$  1.41, 1H, brs) to C-14 ( $\delta$  64.9). The HMBC correlations between H-18 ( $\delta$  1.12, 3H, s) and C-3 ( $\delta$  37.9), C-5 ( $\delta$  56.1), and COOH ( $\delta$  179.0) suggested that the carboxyl group was located at C-4.

The configuration of OH-14 $\beta$  and OH-15 $\alpha$  were determined upon the following NOESY correlations (figure 2) between H-14 and H-13, H-9, OH-15, between H-15 and OH-14, and between H-13 and H-17. The NOESY correlation between H-18 and H-5 indicated that the carboxyl group could be assigned to C-19. Thus, the structure of compound **1** was elucidated as *ent*-14 $\beta$ ,15 $\alpha$ -dihydroxykaur-16-*en*-19-oic acid.

Compound **2** was obtained as colourless flaky crystal. The quasi-molecular ion peak at m/z 355.2043 [M + Na]<sup>+</sup> in the HRESIMS revealed the molecular formula C<sub>20</sub>H<sub>28</sub>O<sub>4</sub>. Comparison of the <sup>1</sup>H and <sup>13</sup>C NMR spectral data of compounds **1** and **2** (table 1) indicated

	1			2	
position	$\delta_{\rm H}$ (mult., J in Hz)	$\delta_{\rm C}$	position	$\delta_{\rm H}$ (mult., <i>J</i> in Hz)	$\delta_{\mathrm{C}}$
1	1.91 (1H, brd, 13.1)	40.4	1	1.94 (1H, brd, 14.2)	39.7
	1.09 (1H, m)			1.08 (1H, m)	
2	1.77 (1H, m)	19.3	2	1.99 (1H, m)	19.9
	1.66 (1H, m)			1.94 (1H, m)	
3	2.01 (1H, brd, 13.1)	37.9	3	2.19 (1H, brd, 13.7)	37.6
	1.02 (1H, m)			1.06 (1H, m)	
4		43.3	4		43.6
5	1.01 (1H, brs)	56.1	5	1.24 (1H, brs)	55.9
6	1.79 (1H, m)	21.7	6	1.83 (1H, m)	18.8
	1.38 (1H, m)			1.49 (1H, m)	
7	1.76 (1H, m)	42.0	7	2.37 (1H, brd, 12.0)	36.6
	1.45 (1H, m)			1.45 (1H, m)	
8		44.8	8		50.5
9	1.41 (1H, brs)	54.8	9	1.40 (1H, brs)	63.1
10		38.0	10		39.0
11	1.36 (1H, m)	39.3	11	1.91 (1H, m)	33.8
	0.82 (1H, m)			1.41 (1H, m)	
12	1.86 (1H, dt, 13.2, 3.5)	36.0	12	2.10 (1H, dt, 13.1, 3.6)	41.1
	0.99 (1H, brd, 13.2)			1.98 (1H, brd, 13.1)	
13	1.51 (1H, m)	39.2	13	2.06 (1H, m)	36.9
14	3.58 (1H, d, 4.7)	64.9	14	4.05 (1H, d, 4.6)	66.3
15	3.76 (1H, s)	81.9	15		209.6
16		158.6	16		150.2
17	4.91 (1H, brs)	105.7	17	5.87 (1H, s)	113.0
	4.89 (1H, brs)			5.27 (1H, s)	
18	1.12 (3H, s)	29.0	18	1.28 (3H, s)	28.9
19	12.03 (1H, s, COOH)	179.0	19	~ • /	181.9
20	0.80 (3H, s)	15.6	20	0.95 (3H, s)	15.6

Table 1. NMR spectral data of compounds 1 and 2 (1 in DMSO- $d_6$ , 2 in CDCl<sub>3</sub>)\*.

\*Assignments were succeeded by HSQC, HMBC, and NOESY experiments.



Figure 2. Important HMBC ( $\rightarrow$ ) and NOESY ( $\leftrightarrow$ ) correlations in compounds 1 and 2.

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that the structure of **2** was almost identical to that of **1**, except that the oxymethine (C-15) in **1** was replaced by the carbonyl group in **2**. Compound **1** was converted to **2** by DDQ [21]. Compound **2** was thus determined to be *ent*-14 $\beta$ -hydroxy-15-oxokaur-16-*en*-19-oic acid, which was confirmed by HSQC, HMBC and NOESY experiments (figure 2).

Compound **3** was obtained as colourless needles. The <sup>1</sup>H and <sup>13</sup>C NMR spectra showed the presence of four tertiary methyls, an *exo*-methylene and three oxygenated methines. The presence of twenty-three signals in <sup>13</sup>C NMR spectrum is consistent with the molecular formula  $C_{23}H_{34}O_5$ , provided by the quasi-molecular ion peak at m/z 413.2043 [M + Na]<sup>+</sup> in the HRESIMS. An isopropylidene group was recognized from the IR absorption bands at  $\nu_{max}1162$  and 1056 cm<sup>-1</sup>, and the <sup>13</sup>C NMR signals at  $\delta$  112.4, 25.6, and 26.0. The above evidence indicated that compound **3** should be an acetonide of an *ent*-kaurane-type diterpene.

The comparison of the <sup>1</sup>H and <sup>13</sup>C NMR data of compound **3** with those of **6** [2] suggested that compound **3** was 11,12-acetonide of **6**. In view of the isolation process, compound **3** may be an artifact of **6**, which was verified by the conversion of **6** in acetone to **3** in the presence of silica gel at room temperature.

#### 3. Experimental

#### 3.1 General experimental procedures

Melting points were measured with an X-6 melting point apparatus and are uncorrected. Optical rotations were measured on a Perkin–Elmer 341 automatic polarimeter. UV and IR spectra were carried out on a Lambda 35 spectrometer and a Perkin–Elmer Spectrum One FT-IR spectrometer (KBr disk), respectively. Mass spectra were obtained on a Finnigan-LCQ<sup>DECA</sup> mass spectrometer (ESIMS) and a Bruker Daltonics Bio-TOF-Q mass spectrometer (HRESIMS). NMR spectra were recorded on a Bruker Avance 600 spectrometer with TMS as internal standard (<sup>1</sup>H: 600 MHz, <sup>13</sup>C: 150 MHz). Column chromatography (CC) was carried out on silica gel (200–300 mesh, Qingdao Haiyang Chemical Group Co. of China), Silica gel 60 (0.015–0.040 mm, Merck), MCI gel (75–150  $\mu$ m, Mitsubishi), and RP-C<sub>18</sub> silica gel (Prepex 40–63  $\mu$ m, Phenomenex). Precoated plates (silica gel GF<sub>254</sub>, 0–40  $\mu$ m) activated at 110°C for 2 h were used for thin-layer chromatography (TLC).

#### 3.2 Plant material

The aerial parts of *N. insignis* were collected in Yanbian County, Panzhihua City of Sichuan Province in May 2004 and identified by Prof. Fading Fu in Chengdu Institute of Biology, the Chinese Academy of Sciences (CAS). A voucher specimen (A-182) is deposited at the Herbarium of Chengdu Institute of Biology, CAS.

### 3.3 Extraction and isolation

The air-dried and powdered aerial part of *N. insignis* (4.5 kg) was soaked with 95% ethanol (25 L  $\times$  3, each 7 d) at room temperature. The solvents were removed under reduced pressure to give 354 g residue, which was suspended in H<sub>2</sub>O (1.5 L) and then extracted

successively with petroleum ether (60–90°C,  $1.5 L \times 4$ ), ethyl acetate ( $1.5 L \times 8$ ) and *n*-butanol ( $1.5 L \times 5$ ) to afford corresponding fractions A (34.3 g), B (56.5 g), and C (114.5 g).

Fraction B (56.5 g) was subjected to CC over silica gel ( $\phi$  8 × L 600 cm, 750 g) eluted with petroleum ether-acetone (50:1, 20:1, 10:1, 5:1, v/v, each 1.5 L) and then eluted with CHCl<sub>3</sub>-MeOH (10:1, 5:1, 3:1, 1:1, v/v, each 1.5 L) to yield subfractions  $B_1$  (0.7 g),  $B_2$  $(0.8 \text{ g}), B_3 (156 \text{ mg}), B_4 (0.8 \text{ g}), B_5 (2.9 \text{ g}), B_6 (5.0 \text{ g}), B_7 (25.2 \text{ g}), and B_8 (14.8 \text{ g}).$ Compounds 1 (22 mg), 4 (57 mg) and 9 (20 mg) were precipitated from B<sub>4</sub>, B<sub>1</sub> and B<sub>2</sub>, respectively. B<sub>3</sub> (156 mg) was separated by silica gel column ( $\phi$  2 × L 25 cm, 20 g) eluted with petroleum ether-acetone (10:1, v/v) to give compounds 13 (15 mg) and 15 (12 mg). Compounds 10 (118 mg), 11 (214 mg) and 25 (23 mg) were obtained from  $B_2$ , which was separated by silica gel column ( $\phi$  2.5 × L 30 cm, 40 g) eluted with petroleum ether-EtOAc (20:1, v/v). B<sub>5</sub> was decolored on an MCI gel column ( $\phi$  3 × L 40 cm, 50 g) eluted with MeOH-H<sub>2</sub>O (8:1, v/v) to yield subfraction  $B_{5(1)}$ , compound 12 (13 mg) was obtained by recrystallizing  $B_{5(1)}$  from methanol.  $B_6$  was subjected to CC on MCI gel ( $\phi$  3 × L 40 cm, 50 g) eluted with MeOH-H<sub>2</sub>O (3:2, v/v) to yield subfractions  $B_{6(1)}$  and  $B_{6(2)}$ . Using CHCl<sub>3</sub>-CH<sub>3</sub>OH (15:1, v/v) as solvent, B<sub>6(1)</sub> was separated by CC over silica gel ( $\phi$  3.5 × L 50 cm, 80 g) to give compounds 3 (12 mg), 14 (35 mg), 16 (65 mg), and 23 (45 mg). A mixture of compounds 6 and 7 (98 mg) was isolated from  $B_{6(2)}$  by silica gel column ( $\phi$  1.5 × L 35 cm, 16 g) eluted with  $CHCl_3$ -CH<sub>3</sub>OH (15:1, v/v).

Fraction A was subjected to CC on MCI gel ( $\phi$  5 × *L* 60 cm, 50 g) eluted with MeOH—H<sub>2</sub>O (3:2, 7:3, 4:1, 9:1, 19:1, 1:0, v/v, each 1.0 L) to yield six subfractions A<sub>1</sub>–A<sub>6</sub>. A<sub>2</sub> (2.8 g) was separated over silica gel column ( $\phi$  2 × *L* 45 cm, 80 g) eluted with petroleum ether–acetone (15:1, v/v) to give compounds **17** (12 mg) and **18** (83 mg). Compounds **2** (6 mg), **5** (53 mg), and **8** (5 mg) were obtained from A<sub>1</sub> (1.3 g) separated by CC on silica gel ( $\phi$  2 × *L* 25 cm, 40 g) eluted with petroleum ether–acetone (10:1, v/v). A<sub>4</sub> (3.2 g) was separated over silica gel column ( $\phi$  3 × *L* 45 cm, 150 g) eluted with petroleum ether-EtOAc (20:1, v/v) to afford compounds **19** (48 mg) and **20** (8 mg). A mixture of compounds **21** and **22** (188 mg) was obtained by separating A<sub>6</sub> over silica gel column ( $\phi$  2 × *L* 40 cm, 90 g) eluted with cyclohexane-EtOAc (50:1, v/v).

Fraction C was subjected to macroporous resin column ( $D_{101}$ , pore size 13–14 nm, 26–60 mesh) to remove sugar by using CH<sub>3</sub>OH—H<sub>2</sub>O (0:1, 7:3, 9:1, 1:0) as solvents to give subfractions C<sub>1</sub>–C<sub>3</sub>. Compound **26** (140 mg) was precipitated from C<sub>3</sub> (20 g) in methanol. Compound **24** (185 mg) was isolated from C<sub>1</sub> (15 g) over a silica gel column ( $\phi$  5 × *L* 60 cm, 500 g) eluted with CHCl<sub>3</sub>—CH<sub>3</sub>OH (5:1, v/v).

#### 3.4 Conversion of compound 1 to compound 2

To the THF solution (5 mL) of compound 1 (10 mg) was added DDQ (21 mg). The mixture was refluxed at 70°C for 5 h and monitored by TLC. Compound 2 was obtained with 86% yield *via* silica gel CC.

#### 3.5 Conversion of compound 6 to compound 3

To the acetone solution (5 mL) of compound **6** (3 mg) was added a small amount of silica gel (15 mg). The mixture was kept at 30°C for 48 h and monitored by TLC. Compound **3** was obtained with 46% yield *via* silica gel CC.

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Figure 3. OPTER diagram of compound 4.

## 3.6 Identification

*Ent*-14 $\beta$ ,15 $\alpha$ -dihydroxykaur-16-*en*-19-oic acid (1): Colourless needles (acetone), m.p. 145–146°C;  $[\alpha]_D^{20}$ -131 (*c* 0.04, CH<sub>3</sub>OH); ESIMS (positive mode) *m/z*: 357 [M + Na]<sup>+</sup>; HRESIMS (positive ion) *m/z*: 357.2043 [M + Na]<sup>+</sup> (calcd for C<sub>20</sub>H<sub>30</sub>NaO<sub>4</sub>, 357.2036); IR $v_{max}$ (KBr)cm<sup>-1</sup>: 3430 (OH), 2936, 1963 (COOH), 2869, 1468, 1053 (C–O–C), 973 (C=CH<sub>2</sub>), 757; <sup>1</sup>H and <sup>13</sup>C NMR spectral data see table 1.

*Ent*-14β-hydroxy-15-oxokaur-16-*en*-19-oic acid (**2**): Colourless flaky crystal (acetone), m.p. 235–236°C;  $[\alpha]_D^{20}$ -140 (*c* 0.10, CH<sub>3</sub>OH); ESIMS (positive mode) *m/z*: 355 [M + Na]<sup>+</sup>; HRESIMS (positive ion) *m/z*: 355.1882 [M + Na]<sup>+</sup> (calcd for C<sub>20</sub>H<sub>28</sub>NaO<sub>4</sub>, 355.1880); IR $v_{max}$ (KBr)cm<sup>-1</sup>: 3478 (OH), 2927, 2871, 1726 (COOH), 1692 (C=O), 1644, 1276, 1179, 1047 (C-O-C), 935 (C=CH<sub>2</sub>), 797; <sup>1</sup>H and <sup>13</sup>C NMR spectral data see table 1.

*Ent*-15α-hydroxykaur-16-*en*-19-oic acid 11,12-acetonide (**3**): Colourless needles (MeOH), m.p. 204–205°C;  $[\alpha]_D^{20}$ -217 (*c* 0.05, CH<sub>3</sub>OH); ESIMS (positive mode) *m/z*: 413 [M + Na]<sup>+</sup>; HRESIMS (positive ion) *m/z*: 413.2310 [M + Na]<sup>+</sup> (calcd for C<sub>23</sub>H<sub>34</sub>NaO<sub>5</sub>, 413.2298); IRv<sub>max</sub>(KBr)cm<sup>-1</sup>: 3400 (OH), 2936, 1723 (COOH), 1467, 1380, 1207, 1162, 1056 (C—O—C), 1027, 866 (C—CH<sub>2</sub>), 780; <sup>1</sup>H NMR (CDCl<sub>3</sub>):  $\delta$  2.18 (1H, d, *J* = 13.4 Hz, H-5), 2.97 (1H, d, *J* = 6.1 Hz, H-9), 4.14 (1H, t, *J* = 6.3 Hz, H-11), 4.11 (1H, t, *J* = 6.5 Hz, H-12), 2.57 (1H, t, *J* = 5.1 Hz, H-13), 3.78 (1H, brd, *J* = 10.0 Hz, H-15), 5.27 and 5.13 (each 1H, brs, H-17), 1.47 (3H, s), 1.31 (3H, s), 1.26 (3H, s), 0.91 (3H,s); <sup>13</sup>C NMR (CDCl<sub>3</sub>):  $\delta$  39.8 (C-1), 18.9 (C-2), 37.6 (C-3), 45.1 (C-4), 55.6 (C-5), 21.1 (C-6), 34.1 (C-7), 44.3 (C-8), 49.2 (C-9), 38.1 (C-10), 73.1 (C-11), 78.0 (C-12), 43.7 (C-13), 38.0 (C-14), 82.0 (C-15), 152.4 (C-16), 107.4 (C-17), 28.8 (C-18), 182.3 (C-19), 15.5 (C-20), 112.4 (C-21), 25.9 (C-22), 25.6 (C-23).

*Ent*-11 $\alpha$ ,16 $\alpha$ -epoxy-15 $\alpha$ -hydroxy-16*S*-kaur-19-oic acid (4): Colourless orthorhombic crystal (acetone), m.p. 264–265°C;  $[\alpha]_D^{20}$ -64 (*c* 0.83, CH<sub>3</sub>OH); ESIMS *m*/*z*: 357 [M + Na]<sup>+</sup>. The IR and NMR data were identical to those reported [2]. The X-ray crystallographic analysis confirmed the structure (figure 3).

X-ray crystallography of **4**: A colourless crystal was obtained from acetone. Crystal data:  $C_{20}H_{30}O_4$ ; Mr = 334.44; dimensions  $0.56 \times 0.36 \times 0.16$  mm; monoclinic, space group  $C_2$ , a = 9.7738(14) Å, b = 11.5661(13) Å, c = 15.559(2) Å,  $\alpha = \beta = \gamma = 90^\circ$ , V = 1758.8(4) Å<sup>3</sup>, Z = 4,  $D_{calc} = 1.263 \text{ g/cm}^3$ ,  $\lambda = 0.71073 \text{ Å}$ ,  $\mu(\text{Mo K}\alpha) = 0.086 \text{ mm}^{-1}$ , F(000) = 728, T = 292 (2) K. Of the 2381 reflections that were collected, 2198 were unique ( $R_{int} = 0.0117$ ); the structure was refined by full-matrix least-squares on F<sup>2</sup>. Final refinement: data/restraints/parameters = 2198/0/229;  $R_1 = 0.0466$  (all data), w $R_2 = 0.0913$  (all data); Absolute structure parameter = 0(10), and GOF = 1.072. The maximum and minimum peaks on the final difference Fourier map corresponded to 0.176 and  $-0.122 \text{ e}^{-}/\text{Å}^{3}$ , respectively. CCDC 6116339 contains the crystallographic data for this paper [22].

Compounds 5-14 and 17-25 were identified by comparing their <sup>1</sup>H- and <sup>13</sup>C-NMR, MS, and IR spectral data with those reported. Compounds **15**, **16**, and **26** were identified by comparing them with authentic samples on TLC and by co-m.p.

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